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## Key indicators

Single-crystal X-ray study  
T = 293 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.012 \text{ \AA}$   
R factor = 0.048  
wR factor = 0.142  
Data-to-parameter ratio = 13.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**(*tert*-Butyl isocyanide)tetracarbonyl[*N*-(1,2,3,4-tetra-phenylbutadienyl)benzophenone imine]diruthenium(I)**

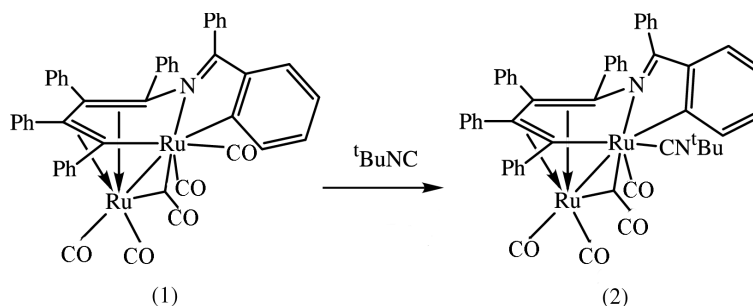
The reaction of  $[\text{Ru}_2\{\mu\text{-PhC}=\text{CPh-CPh}=\text{CPh-N}=\text{CPh}(\text{C}_6\text{H}_4)\}(\mu\text{-CO})(\text{CO})_4]$ , (1), with  ${}^t\text{BuNC}$  leads to the carbonyl substitution product  $[\text{Ru}_2\{\mu\text{-PhC}=\text{CPh-CPh}=\text{CPh-N}=\text{CPh}(\text{C}_6\text{H}_4)\}(\mu\text{-CO})(\text{CO})_3({}^t\text{BuNC})]$  or  $[\text{Ru}_2(\text{C}_5\text{H}_9\text{N})(\text{C}_{41}\text{H}_{29}\text{N})(\text{CO})_4]$ , (2). The X-ray molecular structure of (2) is similar to its parent compound (1) and shows that the isonitrile ligand is in a *cis* arrangement with respect to both the iminic N atom and the C atom of the orthometalated phenyl ring.

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## Comment

This work is part of a project dealing with insertion reactions of unsaturated organic ligands into the C–Ru bonds of polynuclear ruthenium–carbonyl complexes (Cabeza *et al.*, 1998).

The reaction of (1) with *tert*-butylisocyanide does not lead to insertion of the isocyanide into any of the C–Ru bonds of (1) but to the carbonyl substitution product (2). The crystal structure of (2) was determined by X-ray diffraction in order to ascertain the position of the isocyanide ligand in the complex.



The molecular structure of (2) (Fig. 1) is similar to that of its parent compound (1) (Cabeza *et al.*, 1998). The *tert*-butylisocyanide ligand is terminally bound to Ru2 through its C5 atom [ $\text{Ru}2-\text{C}5 = 2.015 (9) \text{ \AA}$  and  $\text{Ru}2-\text{C}5-\text{N}1 = 174.8 (9)^\circ$ ], being *cis* to both the N atom of the imine ligand [ $\text{C}5-\text{Ru}2-\text{N}2 = 94.5 (3)^\circ$ ] and the C atom of the orthometalated phenyl ring [ $\text{C}5-\text{Ru}2-\text{C}12 = 80.4 (3)^\circ$ ].

## Experimental

Synthesis of (2): a solution of (1) (100 mg, 0.114 mmol) and  ${}^t\text{BuNC}$  (26  $\mu\text{l}$ , 0.228 mmol) in dry and deoxygenated tetrahydrofuran (THF; 20 ml) was stirred under nitrogen at reflux temperature for 30 min. The solution was evaporated to dryness under reduced pressure and the residue was separated by column chromatography on neutral alumina (activity IV). Hexane eluted the excess isocyanide. Dichloromethane eluted an orange band which afforded compound (2) upon removal of the solvent. The orange solid was recrystallized at 253 K by liquid diffusion of pentane into a concentrated solution of the

complex in toluene. Yield: 82 mg (77%). Analysis (%), found: C 64.52, H 4.33, N 2.89%; calculated for  $C_{50}H_{38}N_2O_4Ru_2$ : C 64.37, H 4.11, N 3.00%. IR (THF,  $cm^{-1}$ ):  $\nu(CN)$  2166 (*m*);  $\nu(CO)$  2008 (*vs*), 1974 (*s*), 1943 (*m*), 1840 (*m, br*).  $^1H$  NMR ( $CDCl_3$ , 300 MHz, p.p.m.):  $\delta$  7.7–6.5 (*m*, 27H), 5.86 (*d*, 7.4 Hz, 1H), 5.09 (*d*, 7.4 Hz, 1H), 1.56 (*s*, 9H). Compound (1) was prepared as described previously (Cabeza *et al.*, 1998).

#### Crystal data

$[Ru_2(C_5H_9N)(C_{41}H_{29}N)(CO)_4]$   
 $M_r = 932.96$   
 Monoclinic,  $P2_1/c$   
 $a = 20.212$  (10) Å  
 $b = 11.111$  (8) Å  
 $c = 19.362$  (14) Å  
 $\beta = 96.53$  (13)°  
 $V = 4320$  (5) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.434$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 25 reflections  
 $\theta = 10$ –15°  
 $\mu = 0.75$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
 Prismatic, red  
 $0.23 \times 0.20 \times 0.16$  mm

#### Data collection

Nonius CAD-4 diffractometer  
 $\omega$ -2 $\theta$  scans  
 Absorption correction: empirical (*SHELXA*; Sheldrick, 1997a)  
 $T_{min} = 0.840$ ,  $T_{max} = 0.880$   
 7792 measured reflections  
 7562 independent reflections  
 3795 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.072$

$\theta_{max} = 25.0^\circ$   
 $h = -23 \rightarrow 23$   
 $k = 0 \rightarrow 13$   
 $l = 0 \rightarrow 22$   
 3 standard reflections every 200 reflections  
 frequency: 60 min  
 intensity decay: 7.3%

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.142$   
 $S = 1.02$   
 7562 reflections  
 570 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0676P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} < 0.001$   
 $\Delta\rho_{max} = 0.71$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.87$  e Å<sup>-3</sup>

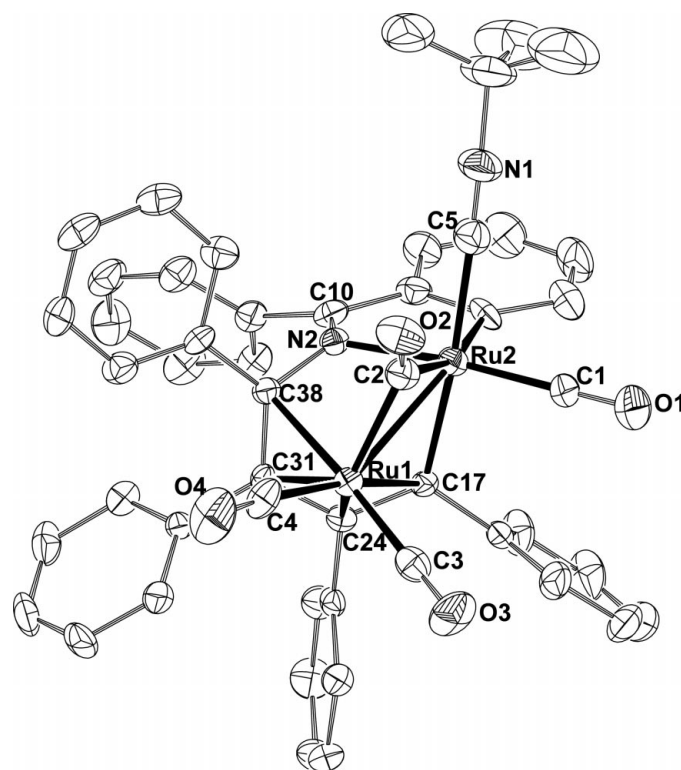
**Table 1**

Selected bond lengths (Å).

Ru1–C3	1.870 (10)	Ru2–C2	2.129 (8)
Ru1–C4	1.890 (8)	O2–C2	1.171 (8)
Ru1–C2	2.054 (7)	O3–C3	1.150 (9)
Ru1–C31	2.215 (7)	O4–C4	1.140 (9)
Ru1–C17	2.260 (6)	O1–C1	1.130 (9)
Ru1–C24	2.313 (7)	N2–C10	1.297 (8)
Ru1–C38	2.317 (7)	N2–C38	1.456 (8)
Ru1–Ru2	2.736 (3)	N1–C5	1.125 (10)
Ru2–C1	1.869 (9)	N1–C6	1.466 (11)
Ru2–N2	2.046 (6)	C17–C24	1.431 (8)
Ru2–C12	2.077 (7)	C24–C31	1.454 (9)
Ru2–C17	2.103 (6)	C31–C38	1.426 (9)

All the H atoms were placed geometrically and refined riding on their parent atoms. At the end of the refinement, the highest peak in the electron density was 1.45 Å from C3 and the deepest hole was 0.03 Å from Ru1.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CRYSDA* (Beurskens *et al.*, 1992); data reduction: *REFLEX* (García-Granda *et al.*, 1999); program(s) used to solve



**Figure 1**

View of a molecule of compound (2). Displacement ellipsoids are shown at the 30% probability level.

structure: *DIRDIF* (Beurskens *et al.*, 1992); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997b); molecular graphics: *PLATON* (Spek, 1982); software used to prepare material for publication: *SHELXL97*.

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