

Cl(2)—Cu—S(2)	94.0 (1)	C(12)—N(5)—C(15)	115 (1)
C(6)—S(1)—C(7)	105.1 (5)		
S(1)—C(6)—C(1)—N(1)	—102.3 (9)	C(1)—C(6)—S(1)—C(7)	61.5 (9)
S(2)—C(11)—C(5)—N(1)	44 (1)	C(5)—C(11)—S(2)—C(12)	83.6 (8)

The structure was solved by direct methods (Gilmore, 1984) and successive Fourier syntheses. Refinement was performed using full-matrix least-squares methods, with non-H atoms anisotropic and H atoms in calculated positions with fixed displacement parameters ($1.2 \times B_{eq}$ of the carrying atom). Calculations were performed with TEXSAN (Molecular Structure Corporation, 1989) software using a VAXstation 3520 computer. The figures were drawn with ORTEPII (Johnson, 1976).

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and bond distances involving H atoms have been deposited with the IUCr (Reference: AB1101). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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[RuCl₂(PH₂Cy)(η⁶-p-cymene)]

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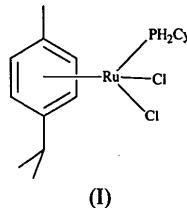
Abstract

The structure of dichloro(cyclohexylphosphine)(η⁶-p-cymene)ruthenium, [RuCl₂(η⁶-C₁₀H₁₄)(C₆H₁₃P)], is reported. The structure determination shows that the C

atoms of the aromatic six-membered ring are almost equidistant from the Ru atom, at distances close to the average value of 2.21 (3) Å. The other metal-ligand distances are Ru—P 2.299 (2), Ru—Cl1 2.411 (1) and Ru—Cl2 2.408 (1) Å. The distances and angles for the Ru environment show a pseudo-octahedral coordination in which the phenyl ring occupies three of the corners of the distorted octahedron.

Comment

This work is part of a project dealing with the reactivity of binuclear ruthenium complexes (Cabeza, Mulla & Riera, 1994), as the title compound, (I), is a derivative of [Ru₂Cl₄(η⁶-p-cymene)₂]. The angles formed by the centroid



(CZ) of the *p*-cymene phenyl ring and the other Ru ligands are CZ—Ru—P 131.1 (2), CZ—Ru—Cl1 128.6 (2) and CZ—Ru—Cl2 128.5 (2)°. These values and the bond distances and angles of the title compound are in good agreement with those found in similar compounds (Regragui, Dixneuf, Taylor & Carty, 1984; Cabeza, Mulla & Riera, 1994).

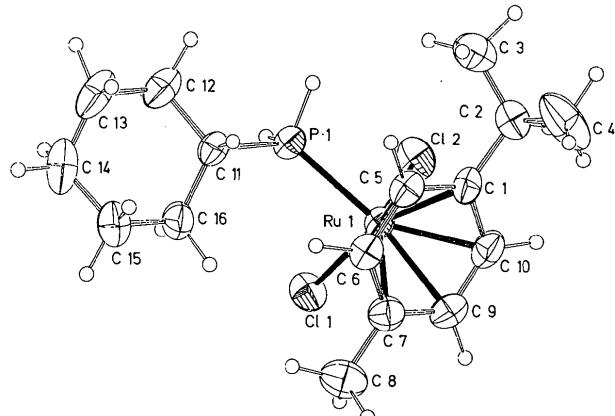


Fig. 1. EUCLID (Spek, 1982) plot of the title compound showing the atomic numbering scheme. Displacement ellipsoids are shown at the 50% probability level.

Experimental

The compound was prepared in quantitative yield from [Ru₂Cl₄(η⁶-p-cymene)₂] and PH₂Cy following the procedure described for the synthesis of the analogous complex [RuCl₂(PPh₂)(η⁶-p-cymene)] (Regragui, Dixneuf, Taylor & Carty, 1984).

Crystal data[RuCl₂(C₁₀H₁₄)(C₆H₁₃P)]*M_r* = 422.32

Orthorhombic

*Pna2*₁*a* = 12.769 (4) Å*b* = 15.826 (10) Å*c* = 9.431 (10) Å*V* = 1905.9 (24) Å³*Z* = 4*D_x* = 1.47 Mg m⁻³*Data collection*Enraf-Nonius CAD-4
diffractometer ω -2θ scans

Absorption correction:

semi-empirical (Walker & Stuart, 1983)

*T*_{min} = 0.97, *T*_{max} = 1.13

3546 measured reflections

3323 independent reflections

2687 observed reflections

[*I* > 2σ(*I*)]*Refinement*Refinement on *F*²*R*(*F*) = 0.0227*wR*(*F*²) = 0.0609*S* = 1.101

3323 reflections

223 parameters

Calculated weights

w = 1/[*σ*²(*F*_o²)
+ (0.0401*P*)²]where *P* = (*F*_o² + 2*F*_c²)/3(*Δ*/*σ*)_{max} = 0.166

Mo *Kα* radiation
 λ = 0.71073 Å
 Cell parameters from 25
 reflections
 θ = 15–18°
 μ = 1.18 mm⁻¹
 T = 293 (2) K
 Transparent prismatic block
 0.40 × 0.17 × 0.10 mm
 Red

*R*_{int} = 0.017
 θ_{max} = 24.91°
 h = 0 → 15
 k = 0 → 18
 l = -11 → 11
 3 standard reflections
 frequency: 60 min
 intensity variation: 2%

$\Delta\rho_{\text{max}}$ = 0.316 e Å⁻³
 $\Delta\rho_{\text{min}}$ = -0.341 e Å⁻³
 Extinction correction: none
 Atomic scattering factors
 from *International Tables for Crystallography* (1992,
 Vol. C, Tables 4.2.6.8 and
 6.1.1.4)

Absolute configuration:
 Flack (1983)

Table 2. Selected geometric parameters (Å, °)

Ru1—C5	2.175 (4)	Ru1—C9	2.238 (5)
Ru1—C1	2.198 (4)	Ru1—P1	2.299 (2)
Ru1—C6	2.207 (3)	Ru1—Cl2	2.408 (1)
Ru1—C7	2.217 (4)	Ru1—Cl1	2.411 (1)
Ru1—C10	2.219 (5)		
C5—Ru1—C1	37.91 (13)	C10—Ru1—C9	36.3 (2)
C5—Ru1—C6	37.76 (13)	P1—Ru1—Cl2	81.18 (5)
C6—Ru1—C7	37.65 (14)	P1—Ru1—Cl1	83.94 (7)
C1—Ru1—C10	37.77 (15)	Cl2—Ru1—Cl1	86.79 (6)
C7—Ru1—C9	37.5 (2)		

Data were collected using profile analysis over all reflections (Lehmann & Larsen, 1974; Grant & Gabe, 1978). The structure was solved by Patterson interpretation using the program *SHELXS90* (Sheldrick, 1990). An isotropic least-squares refinement using a local version of *SHELX76* (Sheldrick, 1976; Van der Maelen Uriá, 1991) converged to *R* = 0.09, then a semi-empirical absorption correction was applied using *DIFABS* (Walker & Stuart, 1983) and the structure was further refined using the program *SHELXL93* (Sheldrick, 1994). Anisotropic refinement followed by a difference Fourier synthesis allowed the location of all of the H atoms. All H atoms were refined isotropically with a common displacement parameter and were left riding with the distances to their parent atoms constrained, except for H101 and H102, which were fixed in the positions located by the difference Fourier synthesis. The final *R* factor in the refinement on *F*² was 0.0227, whereas the final *R* factor in an earlier refinement on *F* was 0.0221. Further geometrical calculations were made with *PARST* (Nardelli, 1983). Data collection: *CAD-4 Software* (Enraf-Nonius, 1989). Cell refinement: *DIRDIF* (Beurskens *et al.*, 1992). Data reduction: *DATAR* (local program). Molecular graphics: *EUCLID* (Spek, 1982). Software used to prepare material for publication: *SHELXL93*.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1074). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Ru1	0.25347 (2)	0.25530 (1)	0.8887†	0.04325 (8)
P1	0.32047 (7)	0.27755 (6)	1.1119 (2)	0.0523 (2)
Cl1	0.12961 (7)	0.16651 (6)	1.0112 (2)	0.0626 (2)
Cl2	0.14528 (7)	0.37288 (6)	0.9580 (2)	0.0689 (3)
C1	0.3330 (3)	0.3279 (2)	0.7217 (4)	0.0548 (8)
C2	0.3495 (3)	0.4228 (2)	0.7048 (5)	0.0684 (10)
C3	0.3970 (4)	0.4649 (3)	0.8280 (6)	0.102 (2)
C4	0.4146 (6)	0.4358 (3)	0.5686 (7)	0.133 (3)
C5	0.4055 (3)	0.2728 (2)	0.7888 (4)	0.0541 (8)
C6	0.3875 (3)	0.1846 (2)	0.8003 (4)	0.0538 (8)
C7	0.2931 (3)	0.1491 (2)	0.7450 (4)	0.0605 (9)
C8	0.2702 (4)	0.0572 (3)	0.7605 (6)	0.0844 (14)
C9	0.2224 (4)	0.2040 (3)	0.6716 (5)	0.0665 (11)
C10	0.2416 (3)	0.2901 (3)	0.6614 (5)	0.0656 (11)
C11	0.4151 (3)	0.2054 (2)	1.1922 (4)	0.0495 (8)
C12	0.4444 (4)	0.2365 (3)	1.3419 (5)	0.0780 (13)
C13	0.5204 (4)	0.1747 (5)	1.4137 (5)	0.094 (2)
C14	0.4820 (4)	0.0853 (4)	1.4130 (5)	0.093 (2)
C15	0.4543 (4)	0.0569 (3)	1.2645 (5)	0.0778 (12)
C16	0.3762 (3)	0.1149 (2)	1.1956 (4)	0.0649 (10)

† Coordinate fixed to define origin.

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